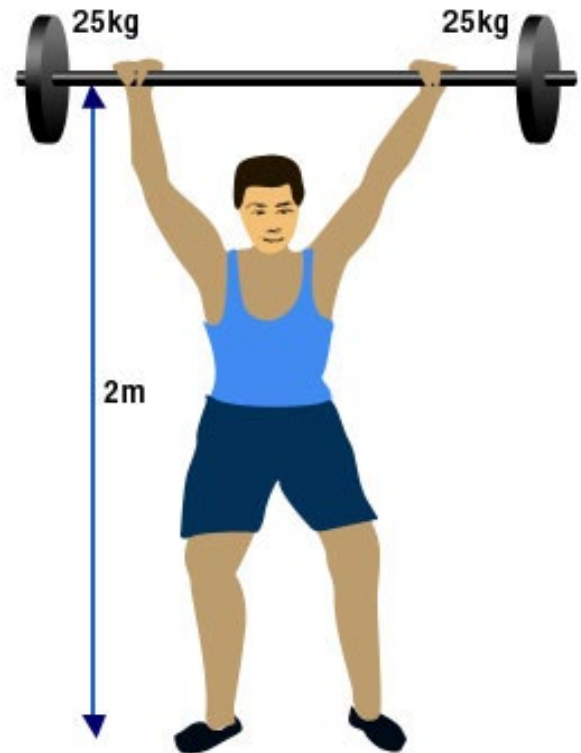


Unit B: Energy Transformations  
**CHAPTER 5: ENERGY  
CONVERSIONS**

# Energy

- Energy is the ability to do **work**
- **Work** is done when a force that is applied to an object moves that object.
- Energy can cause changes to the temperature, shape, speed, or direction of an object



# Energy

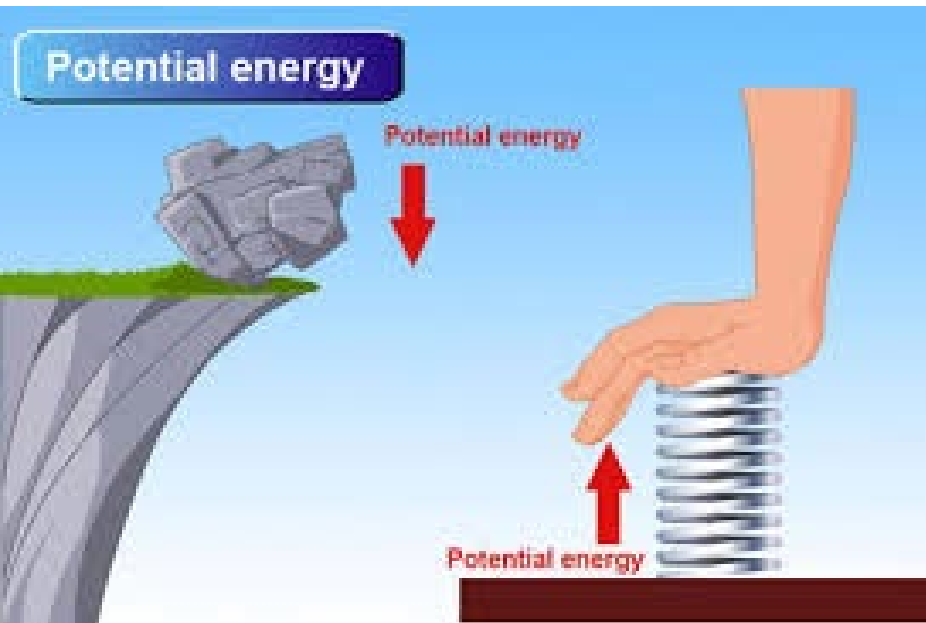


- Where do we get our energy?
- Our main source of energy is food, and if we don't use all the energy it is stored
- Stored energy that has the potential to do work is called **Potential Energy**
- Potential Energy is then converted to **kinetic energy**



# Potential Energy

- In physics, **potential energy** is the **energy** that an object has due to its position
- Often the higher off the ground the object is the higher it's potential energy is



# Kinetic Energy

- In physics, the **kinetic energy** of an object is the **energy** that it possesses due to its motion.
- The body maintains this **kinetic energy** unless its speed changes.

**Kinetic energy**



# Forms of Energy

All forms of energy fall under two categories

## POTENTIAL

Stored energy or energy of position (gravitational)

## KINETIC

Energy of motion (motion of waves, electrons, atoms, molecules, and substances)



Energy in



Energy out



As Kinetic Energy increases  
Potential Energy decreases

So as Kinetic Energy decreases,  
Potential Energy increases!

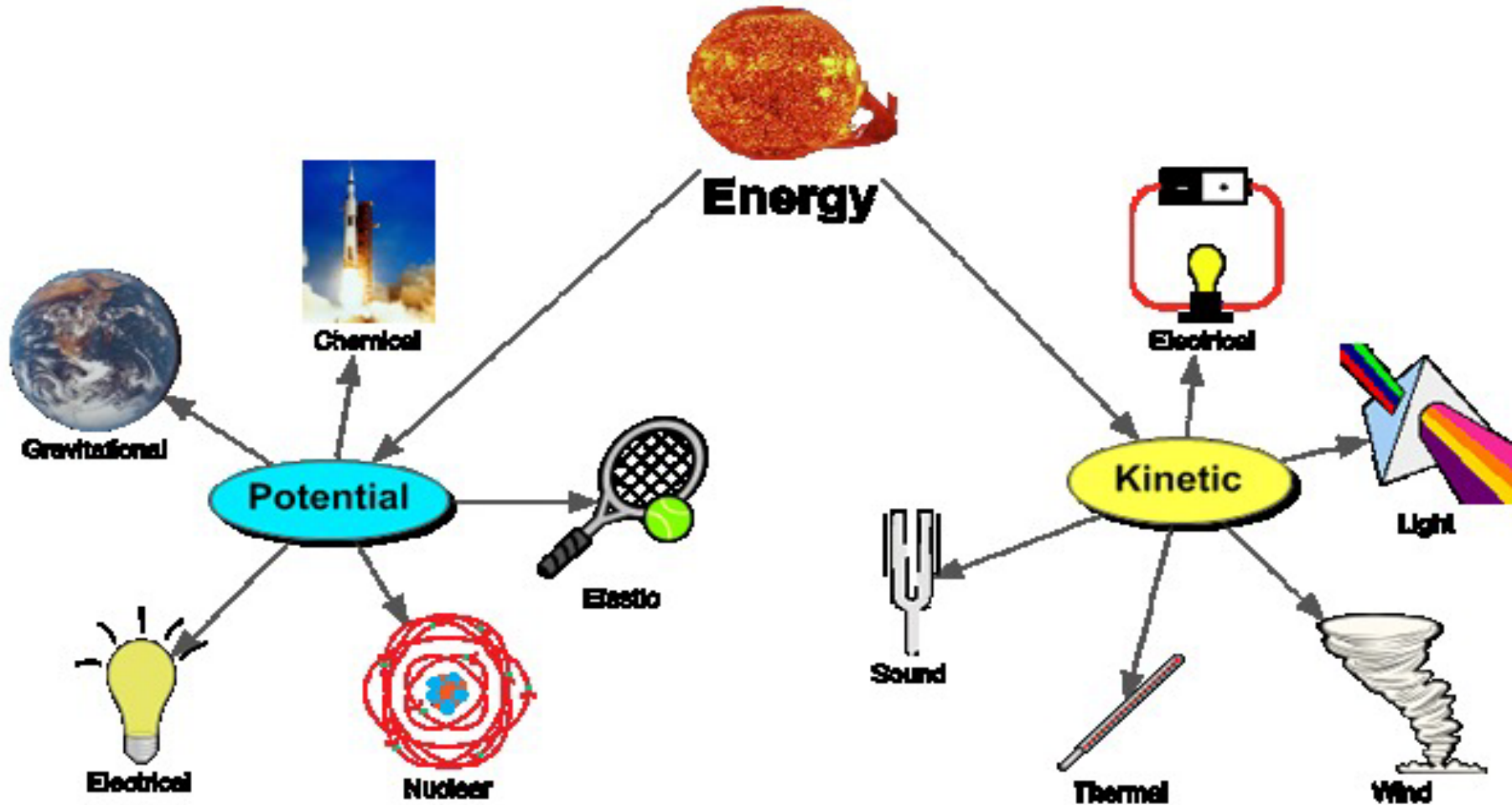


# Many Forms of Energy

- Energy Conversions – turning one form of energy into another
- We probably wouldn't be able to survive without certain energy conversions
- Without energy from the sun, life wouldn't exist on earth
- As with food webs most energy will start with plants



# Forms of Energy





# Examples

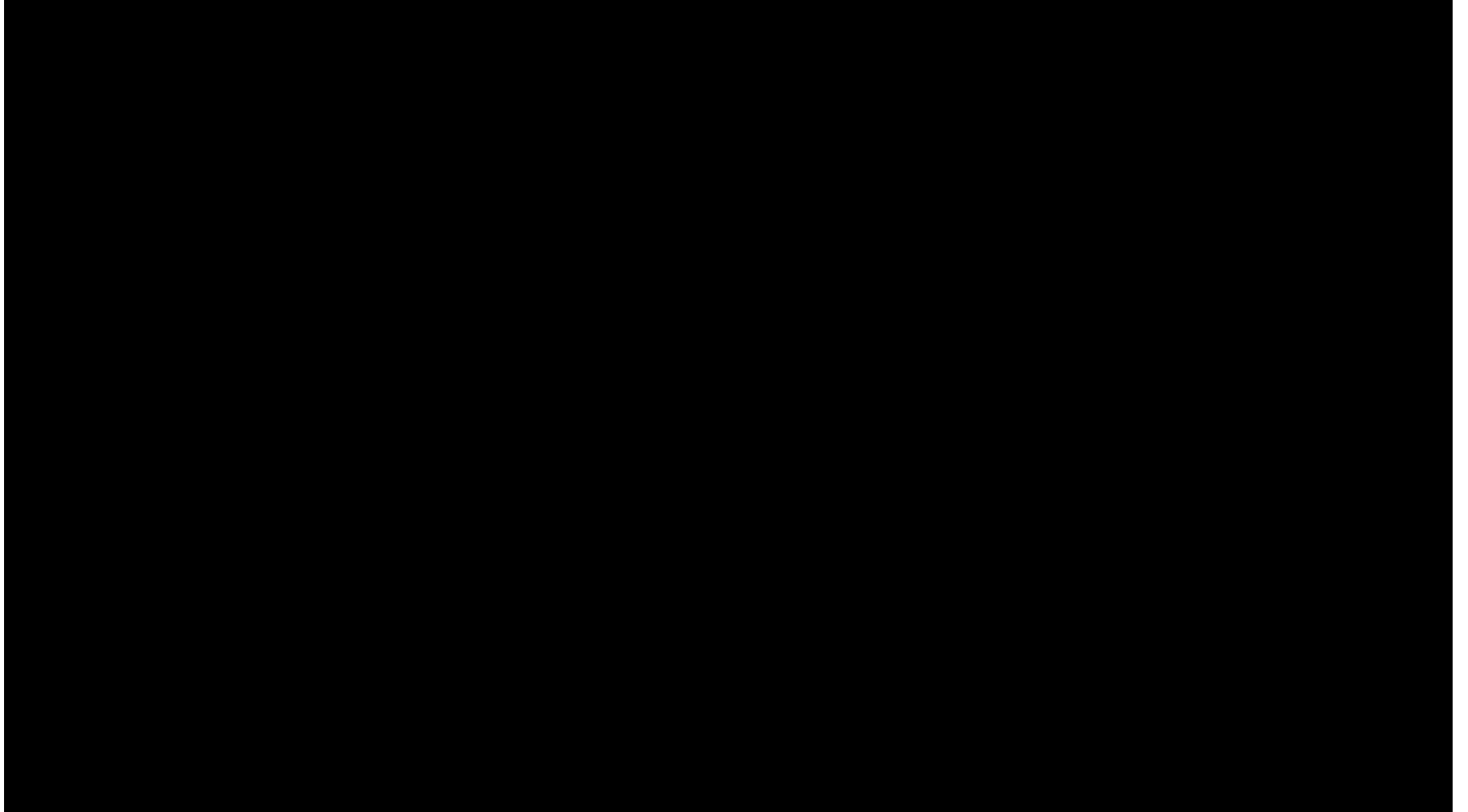
1. Eating food
2. Light bulbs
3. Dropping a beet
4. Gasoline in a car
5. Wind Turbine



1. Chemical  $\rightarrow$  Potential
2. Electrical  $\rightarrow$  Light
3. Potential  $\rightarrow$  Kinetic
4. Chemical  $\rightarrow$  Kinetic
5. Kinetic  $\rightarrow$  Electrical



# SCIENCE!



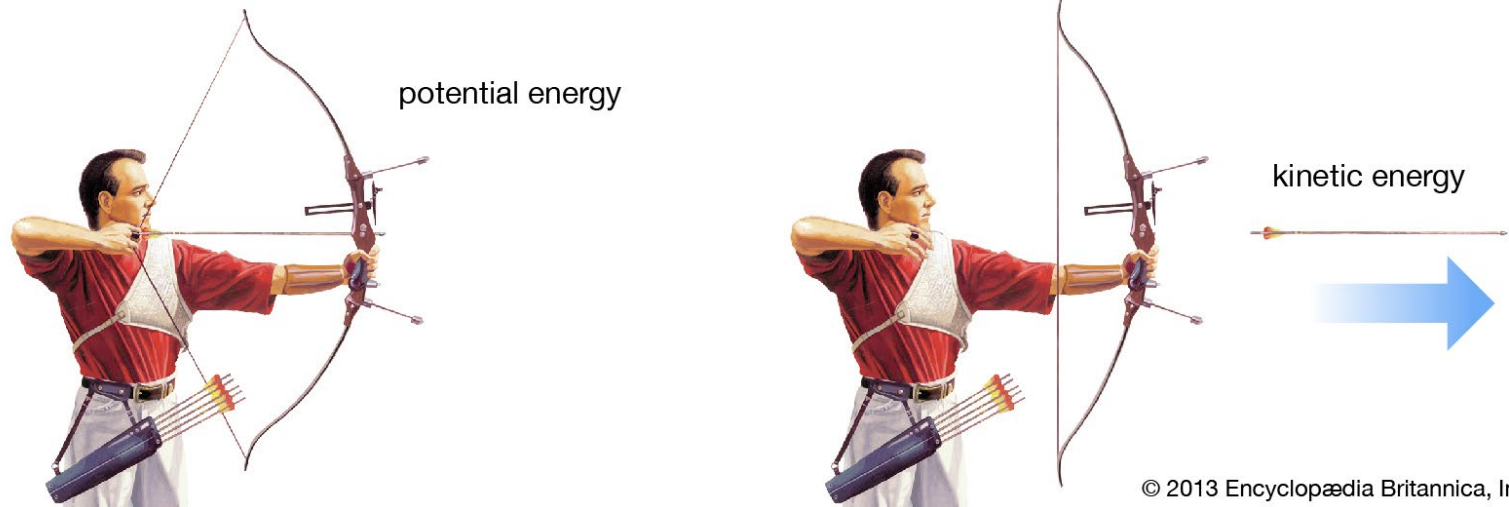
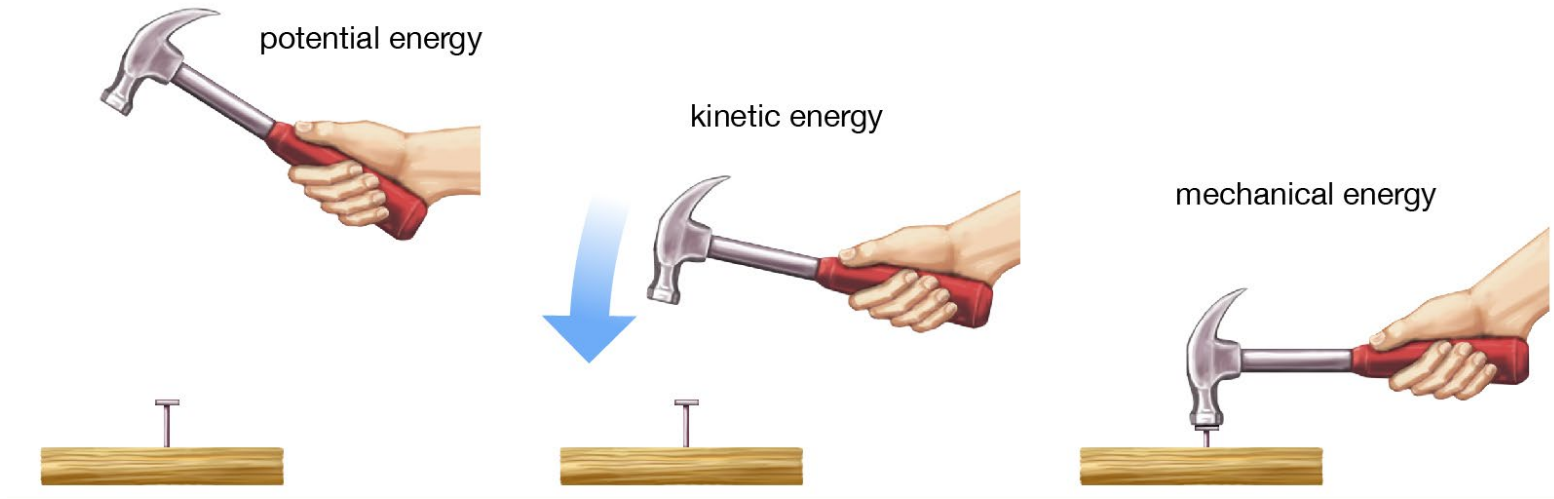
## 2 B Continued



# Chapter 5 continued

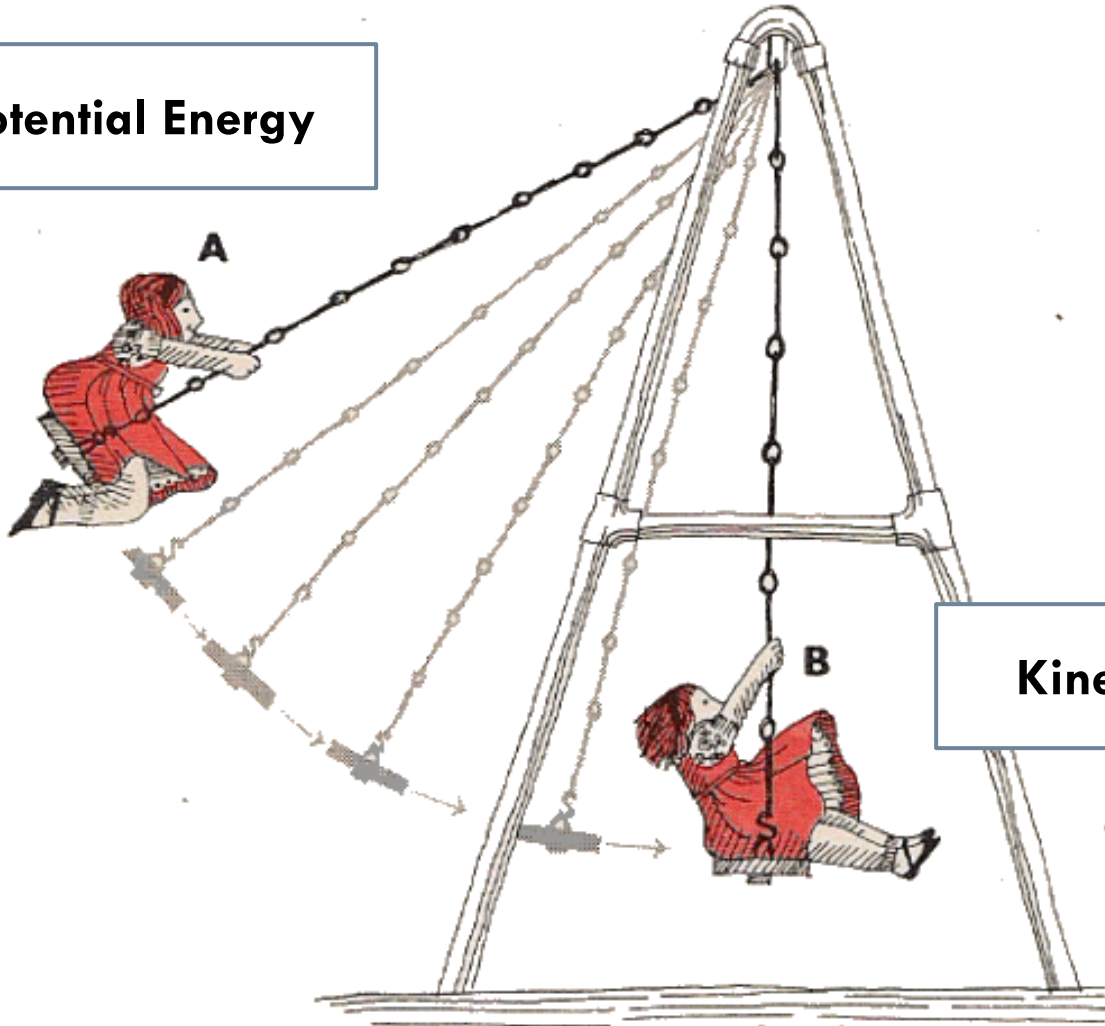


# Review



# Review

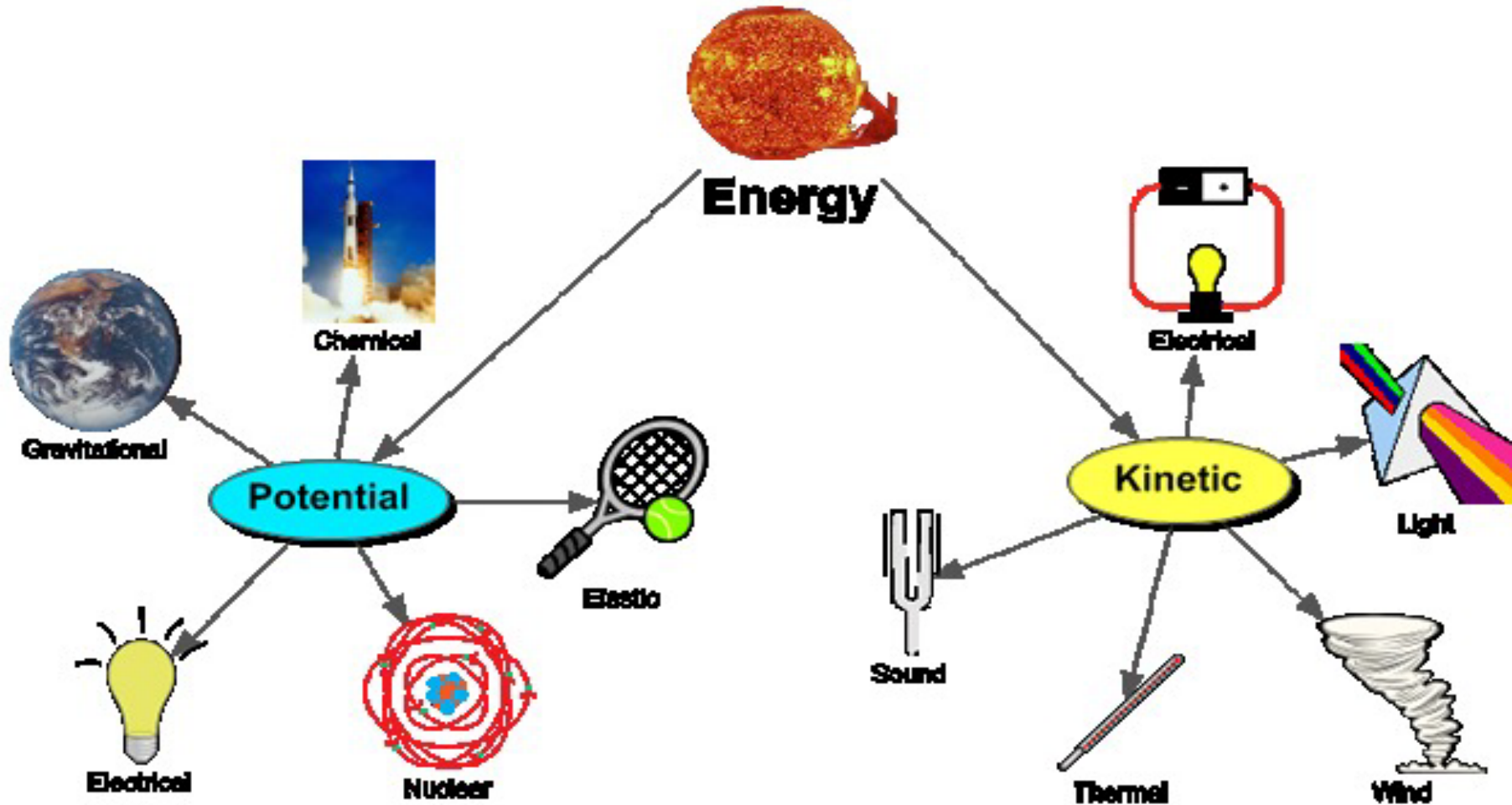
Potential Energy



Kinetic Energy



# Forms of Energy



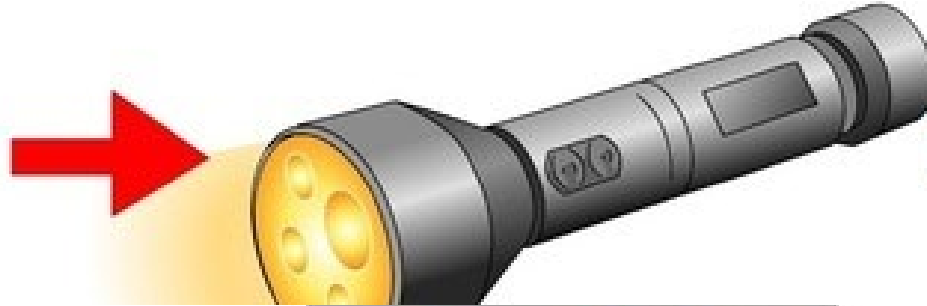
# Energy Conversions

- In order to do work energy constantly changes from one form to another
- **Input energy** – energy that enters the system
- Energy is changed from one form to another at the **converter**
- **Output energy** – energy that leaves the system



# Input → Converter → Output

Chemical  
and  
electrical  
energy



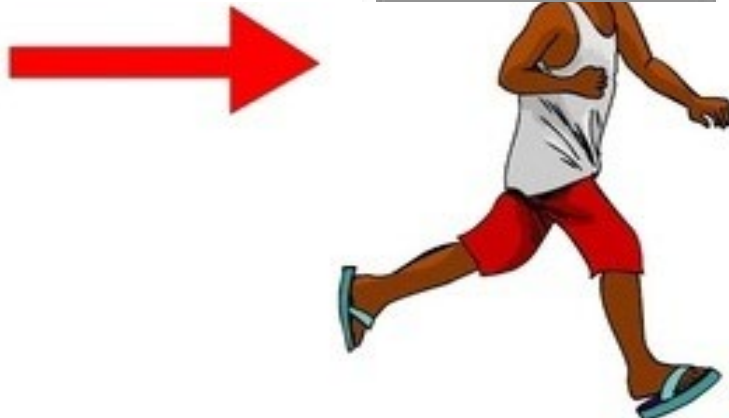
Light  
and  
heat  
energy

Electrical  
energy



Light  
and  
sound  
energy

Chemical  
energy



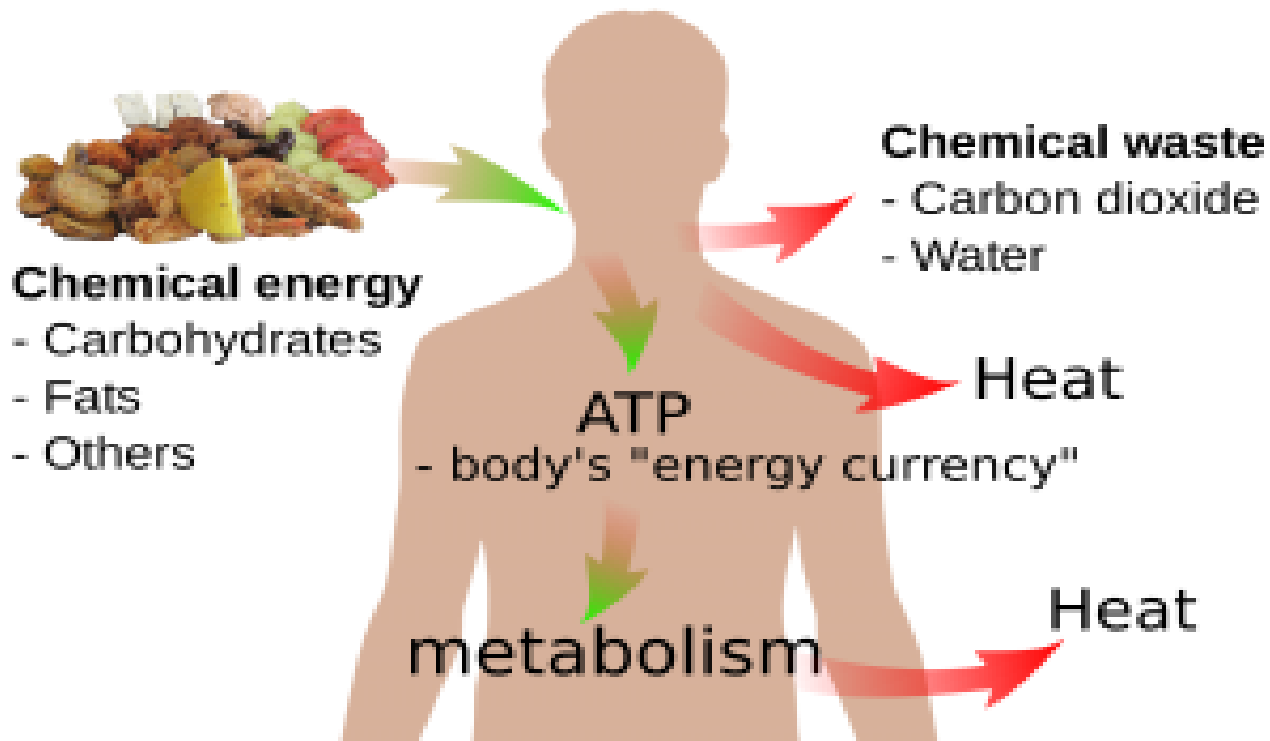
Movement  
and  
heat  
energy

# Chemical Conversions

- The heat you feel during exercise is produced by a chemical reaction within the muscles of your body
- **Potential** Energy from food is converted to **Thermal** Energy
- Potential Chemical Energy → Kinetic Energy
- Potential Chemical Energy → Thermal Energy

# The Body

## Energy and human life



# Thermal Energy Waste

- ❑ Released during most activities – often as waste energy
- ❑ Also produced by machines
- ❑ Usually due to **friction**
- ❑ Opposes motion
- ❑ Excess heat can be dangerous in both man and machine



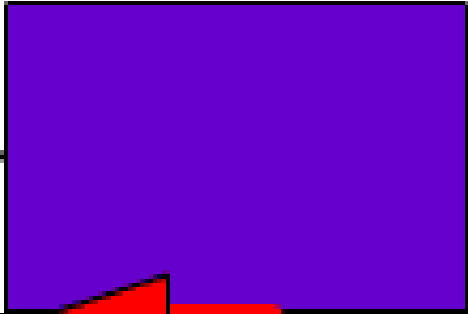




# Friction

Motion →

Pushing force



Friction



# Conversion of Energy

- What types of energy are produced by?



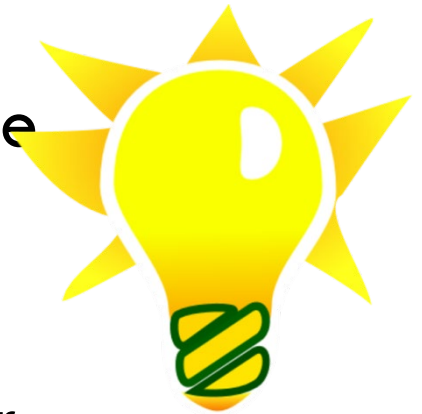
# Conservation of Energy

- Energy cannot be created or destroyed
- **Input Energy = Output Energy**
- iPhone
- Input: Chemical Energy = Output: thermal + kinetic + light + sound
- Energy does not disappear it is just changed to different forms that may be hard to detect

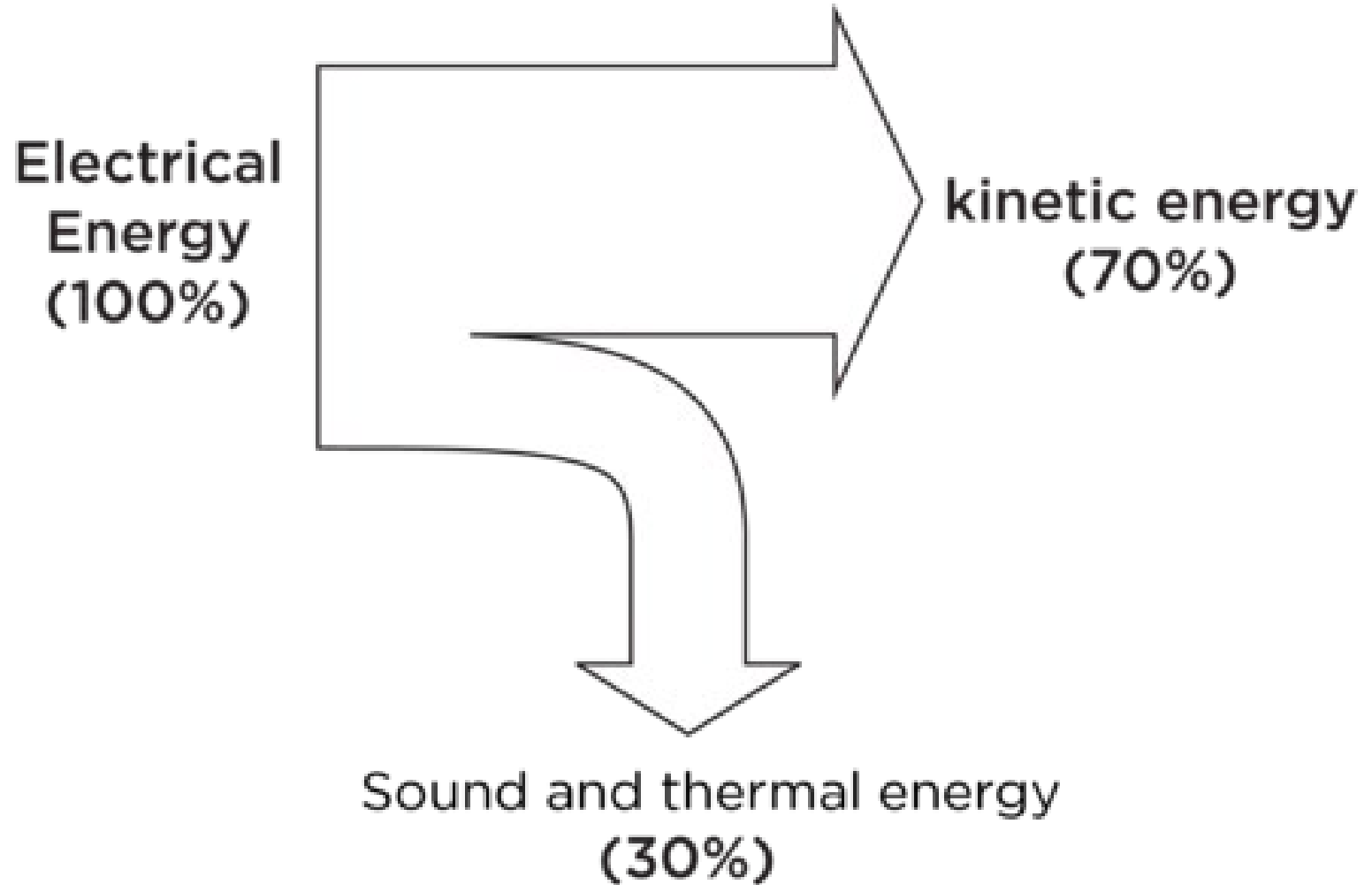


# Energy Conversion System

- **Input Energy = Output Energy**
- But **Input energy** does not equal **useful output energy**
- Every process will give off some sort of waste energy and not be 100% efficient
- In cars only about 1/3 of the energy provided by gasoline is used to move the car
- For lightbulbs 95% of the energy is waste energy in the form of heat



# Example



# End of Chapter 5

